

## Synthesis of Cyclopent[*b*]indoles by Formal [3+2]-Addition of Indolylmethyl Cations to Alkenes

Carrie-Ann Harrison, Ralf Leineweber, Christopher J. Moody\*  
and Jonathan M. J. Williams

Department of Chemistry, Loughborough University of Technology, Loughborough, Leicestershire LE11 3TU, U.K.

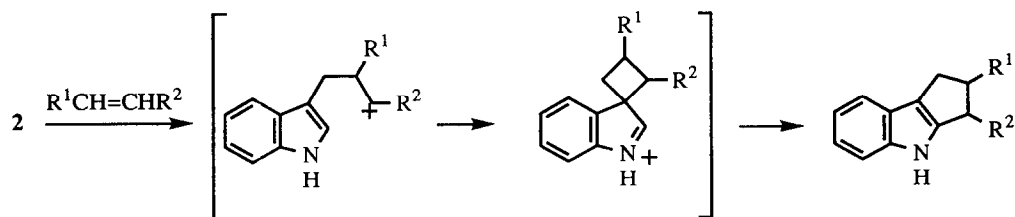
*Abstract:* Treatment of indole-2- or 3-methanols with tin(IV) chloride as Lewis acid in the presence of styrenes results in formal [3+2]-addition of the indole stabilised cation to the alkene to give cyclopent[*b*]indoles with a high degree of stereoselectivity; use of methylcyclohexene as the alkene component gave the *cis*-fused cyclopent[*b*]indole **17**, which was independently synthesised in enantiomerically pure form from the diketone **18**.

The cyclopent[*b*]indole ring system occurs in a number of indole alkaloids, notably the tremorgenic mycotoxins such as paxilline, paspaline, the lolitrems, penitrems and janthitrems,<sup>1</sup> and the monoterpene yuehchukene.<sup>2</sup> We now report a new approach<sup>3</sup> to cyclopent[*b*]indoles based on the formal [3+2]-cycloaddition of the stabilised cation derived from indole-2- or 3-methanols to alkenes.

It is well known that on treatment with acids, indole-3-methanol **1** is readily converted into a stabilised cation **2** which subsequently reacts further to give 3,3'-diindolylmethane.<sup>4</sup> The same product is obtained by reaction of indole itself with formaldehyde,<sup>4</sup> and further examples of 'dimerisation' reactions involving indole stabilised cations (derived from indolemethanols as above or by protonation of vinylindoles) are known.<sup>2a,2d,5</sup>



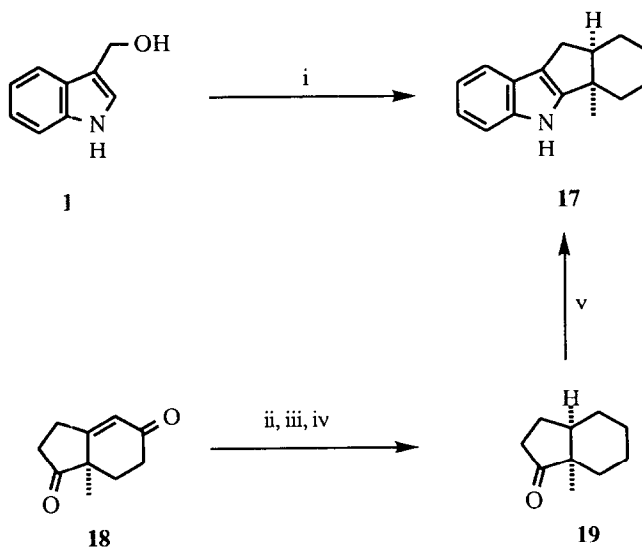
By analogy with the recently reported preparation of dihydroindenes by reaction of benzylic cations with styrenes,<sup>6</sup> it seemed possible that, in the presence of a sufficiently reactive alkene, the 'dimerisation' of indole stabilised cations might be suppressed in favour of a formal [3+2]-addition to the alkene to give cyclopent[*b*]indoles as shown in Scheme 1. This indeed proves to be the case and the reaction provides a simple route to a range of cyclopent[*b*]indoles (Table 1).



Scheme 1

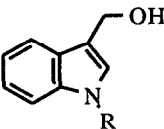

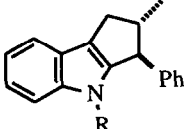
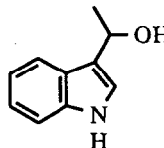

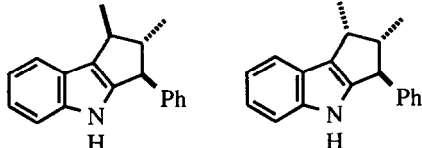

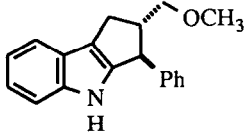
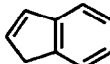
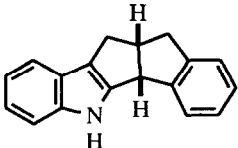
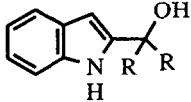
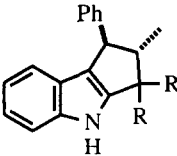
Thus treatment of a mixture of indole-3-methanol **1**, or its *N*-methyl derivative **3**, and  $\beta$ -methylstyrene **4** with tin(IV) chloride at  $-78^\circ\text{C}$  gave, after aqueous work-up and chromatography, the desired cyclopent[b]indoles **5** (55%) and **6** (55%), the *trans*-stereochemistry being proved by NOE difference spectroscopy. The substituted indole-3-methanol **7** reacts similarly with the styrene **4** to give the cyclopent[b]indoles **8** and **9** (63% combined) as a 1:10 mixture of diastereomers. Thus, in common with the related reactions involving benzylic cations,<sup>6</sup> the formal [3+2]-addition reactions of indole stabilised cations with alkenes are highly stereoselective; not only is the original alkene stereochemistry preserved, but the new stereocentre at C-1 is also formed stereoselectively. Angle has speculated about the origin of the stereocentre in his related cationic cyclisations,<sup>6</sup> but the exact mechanism remains unknown, although in the case of our indoles we assume that a 4-membered spiroindolenine must be an intermediate (Scheme 1).<sup>7</sup>

The method was extended to other alkenes, (the allylic ether **10** and indene), and to the indole-2-methanols **13** and **14** to give the corresponding cyclopent[b]indoles (**11**, **12**, **15** and **16**) stereospecifically albeit in poorer yield (Table 1). Finally, the use of alkenes other than styrenes was investigated. Although no cyclopentindoles could be isolated from reactions involving dihydropyran, 1-diethylaminocyclohexene or allyltri-isopropylsilane, the use of 1-methylcyclohexene gave the octahydroindenoindole **17** in 18% yield (Scheme 2). The structure and *cis*-stereochemistry was confirmed by an independent synthesis of enantiomerically pure (-)-**17**, starting from the known *cis*-fused ketone **19**,<sup>8</sup> prepared by reduction<sup>9</sup> of the enantiomerically pure (R)-diketone **18**<sup>10</sup> (Scheme 2).



**Scheme 2.** Reagents: i, 1-methylcyclohexene,  $\text{SnCl}_4$ ,  $\text{CH}_2\text{Cl}_2$ ,  $-78^\circ\text{C}$ ; ii,  $\text{NaBH}_4$ ,  $\text{CF}_3\text{CO}_2\text{H}$ ; iii,  $\text{H}_2$ , Pd-C, EtOH; iv, PCC,  $\text{CH}_2\text{Cl}_2$ ; v,  $\text{PhNHNH}_2$ ,  $\text{H}^+$ .

**Table 1.** Synthesis of cyclop[*b*]indoles by tin(IV) chloride mediated reaction of indolemethanols with alkenes

Indole	Alkene	Product	Yield (%)
 <p><b>1</b> R = H <b>3</b> R = CH<sub>3</sub></p>	 <p><b>4</b> [E:Z = 9:1]</p>	 <p><b>5</b> R = H <b>6</b> R = CH<sub>3</sub></p>	<p><b>5</b> 55 <b>6</b> 55</p>
 <p><b>7</b></p>	<p><b>4</b></p> 	 <p><b>8</b>                      <b>9</b></p>	<p><b>8+9</b> 63 [<b>8:9</b> = 1:10]</p>
<p><b>1</b></p>	 <p><b>10</b></p>	 <p><b>11</b></p>	<p>22</p>
<p><b>1</b></p>		 <p><b>12</b></p>	<p>19</p>
 <p><b>13</b> R = H <b>14</b> R = CH<sub>3</sub></p>	<p><b>4</b></p>	 <p><b>15</b> R = H <b>16</b> R = CH<sub>3</sub></p>	<p><b>15</b> 27 <b>16</b> 17</p>

### General Procedure for Cationic Cyclisation

Tin(IV) chloride (4 equiv) was added slowly to a stirred solution of the indolemethanol (1 equiv) and the alkene (2 equiv; 5 equiv in the case of methylcyclohexene) in dry dichloromethane under nitrogen at  $-78^{\circ}\text{C}$ . After the addition was complete, the mixture was stirred for 10 min at  $-78^{\circ}\text{C}$ , allowed to warm to room temperature, and stirred for a further 40 min (15 h for indolemethanol **13**). The solution was poured into saturated aqueous sodium hydrogen carbonate, and extracted with dichloromethane. The extracts were washed with water, dried ( $\text{MgSO}_4$ ), and evaporated. The residue was purified by flash chromatography on silica gel to give the cyclopent[*b*]indole.

### ACKNOWLEDGEMENTS

We thank the S.E.R.C. for their support of this work, Dr. O. Howarth at the SERC NMR Service at Warwick for 400 MHz NMR spectra, and Professor E. Winterfeldt, Dr G. W. Weaver and Dr. G. Sauer for helpful information concerning the reduction of diketone **18**.

### REFERENCES AND NOTES

- For a review, see: Steyn, P.S.; Vleggar, R. *Fortschr. Chem. Org. Naturst.* **1985**, *48*, 1.
- (a) Cheng, K.-F.; Kong, Y.-C.; Chan, T.-Y. *J. Chem. Soc., Chem. Commun.* **1985**, 48; (b) Wenkert, E.; Moeller, P. D. R.; Pietre, S. R.; McPhail, A. T. *J. Org. Chem.* **1987**, *52*, 3404. (c) Kutney, J. P.; Lopez, F. P.; Huang, S.-P.; Kurobe, H.; Flogans, R.; Piotrowska, K.; Rettig, S. J. *Can. J. Chem.* **1991**, *69*, 949; (d) Sheu, J.-H.; Chen, Y.-K.; Hong, Y.-L. V. *Tetrahedron Lett.* **1991**, *32*, 1045; (e) Bergman, J.; Venemalm, L. *Tetrahedron* **1992**, *48*, 759; (f) Henry, K. J.; Grieco, P. A. *J. Chem. Soc., Chem. Commun.* **1993**, 510.
- For another approach, see: Bergman, J.; Venemalm, L.; Gogoll, A. *Tetrahedron* **1990**, *46*, 6067.
- Sundberg, R. J. *The Chemistry of Indoles*, Academic Press, New York, 1970.
- For example, see: Bergman, J.; Högberg, S.; Lindström, J.-O. *Tetrahedron* **1970**, *26*, 3347; Pfeuffer, L.; Pindur, U. *Helv. Chim. Acta* **1988**, *71*, 467; Bergman, J.; Norrby, P.-O.; Tilstam, U.; Venemalm, L. *Tetrahedron* **1989**, *45*, 5549; Eitel, M.; Pindur, U. *J. Org. Chem.* **1990**, *55*, 5368; Black, D. StC.; Craig, D. C.; Kumar, N. *Tetrahedron Lett.* **1991**, *32*, 1587.
- Angle, S. R.; Arnaiz, D. O. *J. Org. Chem.* **1992**, *57*, 5937.
- The intermediacy of a 4-membered spiroindolenine in the cyclisation reaction of an indole-3-propanol derivative using methanesulfonyl chloride and triethylamine has been demonstrated recently; Ganesan, A.; Heathcock, C. H. *Tetrahedron Lett.* **1993**, *34*, 439.
- Demailly, G.; Solladie, G. *Bull. Soc. Chim. Fr.* **1975**, 2128.
- For another use of sodium borohydride - trifluoroacetic acid in the reduction of ketones, see: Gribble, G. W.; Kelly, W. J.; Emery, S. E. *Synthesis* **1978**, 763.
- Eder, U.; Sauer, G.; Wiechert, R. *Angew. Chem. Int. Ed. Engl.* **1971**, *10*, 496; Rychnovsky, S. D.; Mickus, D. E. *J. Org. Chem.* **1992**, *57*, 2732.

(Received in UK 18 August 1993; accepted 22 October 1993)